

Beeswax

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Beeswax (*Cera alba*) is a natural wax produced by honey bees of the genus *Apis*. The wax is formed into "scales" by eight wax-producing glands in the abdominal segments of worker bees, who discard it in or at the hive. The hive workers collect and use it to form cells for honey-storage and larval and pupal protection within the beehive. Chemically, beeswax consists mainly of esters of fatty acids and various long-chain alcohols.

Beeswax has long-standing applications in human food and flavoring. For example, it is used as a glazing agent, a sweetener, or as a light/heat source. It is edible, in the sense of having similar negligible toxicity to plant waxes, and is approved for food use in most countries and the European Union under the E number E901. However, the wax monoesters in beeswax are poorly hydrolysed in the guts of humans and other mammals, so they have insignificant nutritional value.^[1] Some birds, such as honeyguides, can digest beeswax. Beeswax is the main diet of Wax moth larve.

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Production

The wax is formed by worker bees, which secrete it from eight wax-producing mirror glands on the inner sides of the sternites (the ventral shield or plate of each segment of the body) on abdominal segments 4 to 7.^[2] The sizes of these wax glands depend on the age of the worker, and after many daily flights, these glands begin to gradually atrophy.



A beekeeper from Vojka, Serbia making a bee hive frame.



Commercial honeycomb foundation, made by pressing beeswax between patterned metal rollers



Beeswax cake

The new wax is initially glass-clear and colourless, becoming opaque after mastication and adulteration with pollen by the hive worker bees. Also, the wax becomes progressively more yellow or brown by incorporation of pollen oils and propolis. The wax scales are about 3 mm (0.12 in) across and 0.1 mm (0.0039 in) thick, and about 1100 are required to make a gram of wax.^[3]

Honey bees use the beeswax to build honeycomb cells in which their young are raised with honey and pollen cells being capped for storage. For the wax-making bees to secrete wax, the ambient temperature in the hive must be 33 °C to 36 °C (91 °F to 97 °F).

The amount of honey used by bees to produce wax has not been accurately determined. The book, *Beeswax Production, Harvesting, Processing and Products*, suggests 1 pound of beeswax is used to store 22 pounds honey.^{[4]:41} According to Whitcomb's 1946 experiment, 6.66 to 8.80 pounds of honey yields 1 pound of wax.^{[4]:35} Another study estimated that 24 to 30 pounds of honey is produced per pound of wax.^{[5][6]}

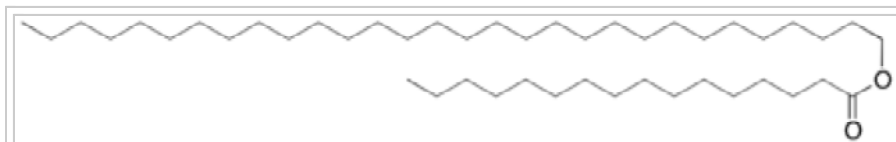
Processing

When beekeepers extract the honey, they cut off the wax caps from each honeycomb cell with an uncapping knife or machine. Its color varies from nearly white to brownish, but most often a shade of yellow, depending on purity, the region, and the type of flowers gathered by the bees. Wax from the brood comb of the honey bee hive tends to be darker than wax from the honeycomb. Impurities accumulate more quickly in the brood comb. Due to the impurities, the wax must be rendered before further use. The leftovers are called slungum.

The wax may be clarified further by heating in water. As with petroleum waxes, it may be softened by dilution with mineral oil or vegetable oil to make it more workable at room temperature.

Physical characteristics

Beeswax is a tough wax formed from a mixture of several compounds.



Triacontanyl palmitate, a wax ester, is a major component of beeswax.

aliphatic alcohols, with the ratio of triacontanyl palmitate $\text{CH}_3(\text{CH}_2)_{29}\text{O}-\text{CO}-(\text{CH}_2)_{14}\text{CH}_3$ to cerotic acid



Uncapping beeswax honeycombs



Fresh wax scales (in the middle of the lower row)

^[8] $\text{CH}_3(\text{CH}_2)_{24}\text{COOH}$, the two principal components, being 6:1.

Beeswax can be classified generally into European and Oriental types. The saponification value is lower (3–5) for European beeswax, and higher (8–9) for Oriental types.

Beeswax has a relatively low melting point range of 62 °C to 64 °C (144 °F to 147 °F). If beeswax is heated above 85 °C (185 °F) discoloration occurs. The flash point of beeswax is 204.4 °C (400 °F).^[9] Density at 15 °C is 958 kg/m³ to 970 kg/m³.

Natural beeswax:^[10] When cold it is brittle; at ordinary temperatures it is tenacious; its fracture is dry and granular. The sp. gr. at 15 °C (59 °F) is from 0.958 to 0.975, that of melted wax at 98 to 99 °C (208.4 to 210.2 °F) compared with water at 15.5 °C (59.9 °F) is 0.822. It softens when held in the hand, and melts at 62 to 66 °C (143.6 to 150.8 °F); it solidifies at 60.5 to 63 °C (140.9 to 145.4 °F).

Uses

Beeswax has many and varied uses. Primarily, it is used by the bees in making their honeycombs. Apart from this use by bees, the use of beeswax has become widespread and varied. Purified and bleached beeswax is used in the production of food, cosmetics, and pharmaceuticals. The three main types of beeswax products are yellow, white, and beeswax absolute. Yellow beeswax is the crude product obtained from the honeycomb, white beeswax is bleached yellow beeswax, and beeswax absolute is yellow beeswax treated with alcohol.^[11] In food preparation, it is used as a coating for cheese; by sealing out the air, protection is given against spoilage (mold growth). Beeswax may also be used as a food additive E901, in small quantities acting as a glazing agent, which serves to prevent water loss, or used to provide surface protection for some fruits. Soft gelatin capsules and tablet coatings may also use E901. Beeswax is also a common ingredient of natural chewing gum.

Use of beeswax in skin care and cosmetics has been increasing. A German study found beeswax to be superior to similar barrier creams (usually mineral oil-based creams such as petroleum jelly), when used according to its protocol.^[12] Beeswax is used in lip balm, lip gloss, hand creams, salves, and moisturizers; and in cosmetics such as eye shadow, blush, and eye liner. Beeswax is also an important ingredient in moustache wax and hair pomades, which make hair look sleek and shiny.

Wax content type	Percentage
Hydrocarbons	14%
Monoesters	35%
Diesters	14%
Triesters	3%
Hydroxy monoesters	4%
Hydroxy polyesters	8%
Acid esters	1%
Acid polyesters	2%
Free fatty acids	12%
Free fatty alcohols	1%
Unidentified	6%








Beeswax candles and figures

Candle-making has long involved the use of beeswax, which is highly flammable, and this material traditionally was prescribed for the making of the Paschal candle or "Easter candle". It is further recommended for the making of other candles used in the liturgy of the Roman Catholic Church.^[13] Beeswax is also the candle constituent of choice in the Orthodox Church.^[14]

^[16] Beeswax is an ingredient in surgical bone wax, which is used during surgery to control bleeding from bone surfaces; shoe polish and furniture polish can both use beeswax as a component, dissolved in turpentine or sometimes blended with linseed oil or tung oil; modeling waxes can also use beeswax as a component; pure beeswax can also be used as an organic surfboard wax.

^[17] Beeswax blended with pine rosin, can serve as an adhesive to attach reed plates to the structure inside a squeezebox. It can also be used to make Cutler's resin, an adhesive used to glue handles onto cutlery knives. It is used in Eastern Europe in egg decoration; it is used for writing, via resist dyeing, on batik eggs (as in pysanky) and for making beaded eggs. Beeswax is used by percussionists to make a surface on tambourines for thumb rolls. It can also be used as a metal injection moulding binder component along with other polymeric binder materials.^[18] Beeswax was formerly used in the manufacture of phonograph cylinders. It may still be used to seal formal legal or Royal decree and academic parchments such as placing an awarding stamp imprimatur of the university upon completion of post-graduate degrees.

 India	23 000
 Ethiopia	5 000
 Argentina	4 700
 Turkey	4 235
 Republic of Korea	3 063
World total	
Source: UN FAOSTAT (http://faostat.fao.org/site/339/default.aspx) ^[15]	

Historical uses

Beeswax was among the first plastics to be used, alongside other natural polymers such as gutta-percha, horn, tortoiseshell, and shellac. For thousands of years, beeswax has had a wide variety of applications; it has been found in the tombs of Egypt, in wrecked Viking ships, and in Roman ruins. Beeswax never goes bad and can be heated and reused. Historically, it has been used:

- As candles - the oldest intact beeswax candles north of the Alps were found in the Alamannic graveyard of Oberflacht, Germany, dating to 6th/7th century AD
- In the manufacture of cosmetics
- As a modelling material in the lost-wax casting process, or *cire perdue*^[19]
- For wax tablets used for a variety of writing purposes
- In encaustic paintings such as the Fayum mummy portraits^[20]
- In bow making
- To strengthen and preserve sewing thread, cordage, shoe laces, etc.



Beeswax candles, Alamannic graveyard (Oberflacht, Germany), 6th/7th c. AD

- As a component of sealing wax
- To strengthen and to forestall splitting and cracking of wind instrument reeds
- To form the mouthpieces of a didgeridoo, and the frets on the Philippine *kutiyapi* – a type of boat lute
- As a sealant or lubricant for bullets in cap and ball firearms
- To stabilize the military explosive Torpex – before being replaced by a petroleum-based product
- In producing Javanese *batik*^[21]
- As an ancient form of dental tooth filling^{[22][23]}



Beeswax as Neolithic Dental Filling

See also

- Carnauba wax
- Honeycomb
- Paraffin wax
- Pysanka

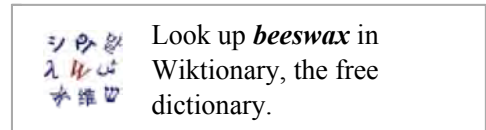
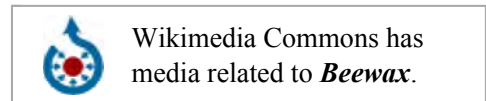
References

1. Beeswax absorption and toxicity (<http://www.efsa.europa.eu/en/efsajournal/doc/615.pdf>). Large amounts of such waxes in the diet pose theoretical toxicological problems for mammals.
2. Sanford, M.T.; Dietz, A. (1976). "The fine structure of the wax gland of the honey bee (*Apis mellifera* L.)". *Apidologie*. **7**: 197–207. doi:10.1051/apido:19760301.
3. Brown, R, H. (1981) *Beeswax* (2nd edition) Bee Books New and Old, Burrowbridge, Somerset UK. ISBN 0-905652-15-0
4. *Beeswax Production, Harvesting, Processing and Products*, Coggshall and Morse. Wicwas Press. 1984-06-01. ISBN 1878075063.
5. Les Crowder (2012-08-31). *Top-Bar Beekeeping: Organic Practices for Honeybee Health*. Chelsea Green Publishing. ISBN 1603584617.
6. Top-bar beekeeping in America (<http://www.beesfordevelopment.org/what-we-do/information-and-education/info/file/1327>).
7. Umney, Nick; Shayne Rivers (2003). *Conservation of Furniture*. Butterworth-Heinemann. p. 164.
8. "LIPID MAPS Databases : LIPID MAPS Lipidomics Gateway". Lipidmaps.org. Retrieved 2013-07-05.
9. "MSDS for beeswax".. No reported autoignition temperature has been reported
10. A Dictionary of Applied Chemistry, Vol. 5. Sir Edward Thorpe. Revised and enlarged edition. Longmans, Green, and Co., London, 1916. "Waxes, Animal and vegetable. Beeswax", p. 737
11. [1] (http://www.webmd.com/vitamins_supplements)
12. Peter J. Frosch; Detlef Peiler; Veit Grunert; Beate Grunenber (July 2003). "Wirksamkeit von Hautschutzprodukten im Vergleich zu Hautpflegeprodukten bei Zahntechnikern – eine kontrollierte Feldstudie. Efficacy of barrier creams in comparison to skin care products in dental laboratory technicians – a controlled trial.". *Journal der Deutschen Dermatologischen Gesellschaft* (in German). Blackwell Synergy. **1** (7): 547–557. doi:10.1046/j.1439-0353.2003.03701.x. PMID 16295040. Retrieved 1/12/2008. "CONCLUSIONS: The results demonstrate that the use of after work moisturizers is highly beneficial and under the chosen study conditions even superior to barrier creams applied at work. This approach is more practical for many professions and may effectively reduce the frequency of irritant contact dermatitis." Check date values in: |access-date= (help)
13. 'Altar Candles" (<http://www.newadvent.org/cathen/01347a.htm>), 1913 Catholic Encyclopedia
14. [2] (<http://www.johnsanidopoulos.com/2010/02/use-of-candles-in-orthodox-church.html>), Use of Candles in the Orthodox Church

15. "Statistics from: Food And Agricultural Organization of United Nations: Economic And Social Department: The Statistical Division". UN Food and Agriculture Organization Corporate Statistical Database. Archived from the original on 2011-07-13.
16. Uwe Wolfmeier, Hans Schmidt, Franz-Leo Heinrichs, Georg Michalczyk, Wolfgang Payer, Wolfram Dietsche, Klaus Boehlke, Gerd Hohner, Josef Wildgruber "Waxes" in Ullmann's Encyclopedia of Industrial Chemistry, Wiley-VCH, Weinheim, 2002. doi:10.1002/14356007.a28_103 (https://dx.doi.org/10.1002%2F14356007.a28_103).
17. "Raw Beeswax Uses" (<http://www.morenature.com/raw-beeswax-uses.html>), MoreNature
18. "Metal Injection Molding Process (MIM)" (<http://www.encyclopedia.com/chemical-engineering-encyclopedia/dictionary/chemical-processes/1793-metal-injection-molding-process-mim>), 2012 EngPedia
19. Congdon, L. O. K. (1985). "Water-Casting Concave-Convex Wax Models for Cire Perdue Bronze Mirrors". *American Journal of Archaeology*. **89** (3): 511–515. JSTOR 504365.
20. Egyptology online (http://www.egyptologyonline.com/mummy_portraits.htm)
21. Ormeling, F. J. 1956. *The Timor problem: a geographical interpretation of an underdeveloped island*. Groningen and The Hague: J. B. Wolters and Martinus Nijhoff.
22. "Oldest tooth filling may have been found – Light Years – CNN.com Blogs". Lightyears.blogs.cnn.com. Retrieved 2013-07-05.
23. "Don't Use Your Teeth". Retrieved 2013-12-13.

External links

- The chemistry of bees (<http://www.chm.bris.ac.uk/webprojects2001/loveridge/>) Joel Loveridge, School of Chemistry, University of Bristol [accessed November 2005]



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